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object has because of its
motion. kinetic theory.

All matter consists of
tiny particles that are in
constant motion.

molecules/atom. the

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particles in a gas usually
consist of. insignificant.

The particles in a gas
are considered to be
small, hard spheres with
an _____ volume.
empty space ...

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energy. The energy an object has because of its motion. kinetic theory .

All matter consists of tiny particles that are in constant motion.

molecules/atom. the particles in a gas usually consist of. empty space.

what is between particles in a gas? rapid, constant, and random.

the motion of the particles in a ...

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earth metals. halogen.
cation. any metal in
Group 1A of the
periodic table. any
metal in Group 2A of

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the periodic table. a
nonmetal in Group 7A
of the periodic table.
any atom or group of
atoms with a positive
charge . alkali metals.
any metal in Group 1A
of the periodic table ...

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because of its motion.

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describe the

assumptions of kinetic

theory as applied to

gases-The particles in a

gas are considered to be

small, hard speres with

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an insignificant volume-
The motion of the
particles in a gas is rapid,
constant, and random ...

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Barometer. Two or more different molecular forms of the same element in t... A solid without an ordered internal structure. The result from the collisions of atoms and molecules in air w... A device that is used to measure atmospheric pressure. Allotrope. Two or more ...

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energy. kinetic theory.

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insignificant. The

energy an object has

because of its motion.

All matter consists of

tiny particles that are in

constant mot... the

particles in a gas usually

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consist of. The particles in a gas are considered to be small, hard sphere... kinetic energy. The energy an object has because of its motion. kinetic ...

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Kinetic Energy. The energy an object has because of its motion.

Kinetic Theory . The theory that states that all matter consists of tiny

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particles that are in constant motion. When applied to gases makes the following fundamental assumptions: The particles in a gas are small hard spheres with insignificant volume, The motion of gas particles is ...

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don't have such hobby?

So, it's important for
you to begin having that
hobby. You know,
reading is not the force.

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reading will guide you to
belong to in greater
than before concept of
life. Reading will be a
clear protest to reach

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â € ¢ Determine the
number of valence ...

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stable electron structure
of a noble gas by losing
electrons. ... Answer the
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